In situ Accelerator-Based Characterization of CaO/Sr/Pd Samples under Deuterium Permeation

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Deuterium permeation through Pd/(CaO+Pd)/Pd sample induced nuclear transmutations;

¹³³Cs \rightarrow ¹⁴¹Pr, ⁸⁸Sr \rightarrow ⁹⁶Mo, ¹³⁸Ba \rightarrow ¹⁵⁰Sm and ¹³⁷Ba \rightarrow ¹⁴⁹Sm.

- 1. Y. Iwamura, M. Sakano and T. Itoh; Jpn. J. Appl. Phys. 41 (2002) 4642-4650.
- 2. Y. Iwamura, T. Itoh, M. Sakano, S. Kuribayashi, Y. Terada, T. Ishikawa and J. Kasagi; Proc. ICCF11, 2004, Marseilles, France.

In the present work;

- The objective is
 - to investigate the dependence of the transformation rate on material, D flux, flow direction, temperature, etc.
 - to find the optimum sample structure.
 - from a point of view of enhancing the transformation yield.

Diagnostic methods;

in situ and simultaneous measurements of compositional change during D permeation by accelerator analyses;

PIXE (Particle induced X-ray emission analysis)

RBS (Rutherford backscattering spectroscopy)

NRA (Nuclear reaction analysis)

ERDA (Elastic recoil detection analysis)

as well as conventional XPS and Desorption measurements.

Experimental procedure; in the present work, X = Sr, Y = CaO







Deuterium permeates through the CaO(~10nm)/Sr/Pd sample from the rear surface out to vacuum.

(1) Depositing Sr on Pd ; (A) electrochemical method



Sr atoms are deposited on one side of a 0.1-mm-thick Pd foil as contamination with an areal density of the order of 10¹⁴ cm⁻².

(1) Depositing Sr on Pd ; (B) ion beam sputtering



(2) Forming a layer of CaO on Sr/Pd ; RF sputtering deposition

RF sputtering condition

- Cathode (sputter target); CaO
- Anode (sample); Sr/Pd
- Ar pressure; 0.1 Torr
- RF power; 100 W
- Plasma exposure time; 5 30 min.
- Sample temperature; 450 K
- CaO deposition rate; 0.1 0.5 nm/min

(2D) XPS characterization after forming the layer of CaO on Sr/Pd



Change in the sample structure is reflected in XPS spectra after repetitive etching of the CaO/Sr/Pd surface.

- The CaO layer thickness x is calculated from the intensity ratio of these peaks, $Y_{\text{Ca-2p}}/Y_{\text{Pd-3d}}$;

$$\exp\left(\frac{x}{\lambda_{\text{CaO-Pd}}}\right) \cdot \left(1 - \exp\left(-\frac{x}{\lambda_{\text{CaO-Ca}}}\right)\right) = \frac{n_{\text{Pd}} \cdot \sigma_{\text{Pd}} \cdot \lambda_{\text{Pd-Pd}} \cdot Y_{\text{Ca-2p}_{2}^{3}/2}}{n_{\text{Ca}} \cdot \sigma_{\text{Ca}} \cdot \lambda_{\text{CaO-Ca}} \cdot Y_{\text{Pd-3d}_{2}^{5}/2}}$$

where

$$Y_{\text{Ca}_2\text{p}_2^3/2} = \phi_{\text{p}} \cdot S \cdot n_{\text{Ca}} \frac{d\sigma_{\text{Ca}}}{d\Omega} \Delta \Omega \cdot \int_0^x \exp\left(-\frac{t}{\lambda_{\text{CaO-Ca}}}\right) dt,$$

$$Y_{\text{Pd-3d}\frac{5}{2}} = \phi_{\text{p}} \cdot S \cdot n_{\text{Pd}} \frac{d\sigma_{\text{Pd}}}{d\Omega} \Delta \Omega \cdot \exp\left(-\frac{x}{\lambda_{\text{CaO-Pd}}}\right) \int_{0}^{\infty} \exp\left(-\frac{t}{\lambda_{\text{Pd-Pd}}}\right) dt.$$

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(2D) XPS characterization after forming the layer of CaO on Sr/Pd



The thickness of the CaO layer has been calculated from the XPS yield of the Pd-3d peak relative to that of Ca-2p to be about 10 nm for the sample #4.

(3) Permeation of D through CaO/Sr/Pd (the sample B)



During D permeation, pressure in the reservoir has been monitored to calculate the number of absorbed/transmitted D atoms.

(3D) NRA during permeation of D through CaO/Sr/Pd (the sample B)



NRA using the D(³He, ⁴He)p reaction

The NRA revealed the almost uniform composition of PdD_{0.65} over the depth region of 0-1.8 μm.

(3D) PIXE analysis during permeation of D through CaO/Sr/Pd (the sample B)



The PIXE spectra revealed existence of a variety of impurities. The Mo- $K_{\alpha/\beta}$ seems to be interfered by unidentified broad peaks.

(3D) PIXE analysis during permeation of D through CaO/Sr/Pd (the sample B)



Variation of the areal density of Sr during permeation of D for 590 hours. Results of ex-situ XPS are also plotted for comparison.

(4) **Desorption of D**₂



Outgassing of the sample after finishing D permeation was necessary for introduction into the XPS chamber. Total amount of desorbed D atoms measured with QMS indicates D/Pd~0.3.

- The process of deuterium charging is described by a 1-dimensional solution of the diffusion equation for the sample with a thickness and area of *a* and *S*, respectively.
- One side of the sample is assumed to be opaque for deuterium, while the other facing to hydrogen gas to give a boundary condition for the deuterium density; $n(a) = n_0$.
- The number of deuterium atoms absorbed in the sample, $N_a(t)$, is given as a function of time by

$$N_{a}(t) = aSn_{0} \left[1 - \sum_{s=0}^{\infty} \left\{ \frac{8}{(2s+1)^{2} \pi^{2}} \cdot \exp\left(-\left(\frac{(2s+1)\pi}{2a}\right)^{2} Dt\right) \right\} \right]$$

- A characteristic time $t_{1/2}$ for the number of deuterium atoms absorbed to reach a half of the saturation value, *i.e.*, $N_{\rm a}(t_{1/2})/aS = N_{\rm a}(\infty)/2aS = 2.9 \times 10^{22}$ cm⁻³, is therefore given by

$$t_{1/2} = \frac{0.1967a^2}{D} = 6 \times 10^1 \,\mathrm{s}$$
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XPS spectra before and after D permeation. Increase in Mo-3d peaks in exchange for decrease in Sr-3d peaks is observed.



XPS analysis implies nuclear transmutation near the CaO/Pd boundary: Sr atoms with areal density of 1.3×10¹⁴ cm⁻² appear to be transformed to Mo atoms of 6.6×10¹³ cm⁻² by D permeation.

(4D) XPS characterization of the sample B after D₂ Desorption



An order of magnitude increase in areal density of Sr (~2 monolayers) was realized by RF sputtering method. However, nuclear transmutation was not recognized for this sample.

Summary of changes in areal densities

	PIXE			XPS	
	Sr	Мо	Others	Sr	Мо
A (electroplating)				-1.3	+0.66
B (I. B. sputtering)	-2.5	<<+ 1.9	<<+1	-15	<<+0.11
	areal densities in 1E14 cm ⁻²				

Sample A: Transmutation efficiency was ~50%. Sample B: Decrease in areal density of Sr was not followed by consistent increase in other elements observable.

Summary

- Implication of nuclear transmutation Sr →Mo has been obtained in a system [vacuum/CaO/Sr/PdD_x/D₂], which is a little simpler than that used by Iwamura *et al*.
- (2) The diagnostic method used to identify the elements was conventional XPS, giving the areal densities of 1.3 ×10¹⁴ cm⁻² (Sr) and 6.6×10¹³ cm⁻² (Mo).
- (3) Extended analytical methods have been prepared; *in situ* and simultaneous PIXE, RBS and NRA/ERDA for areal densities of transmutation elements and deuterium distribution.
- (4) Increase in the areal density of Sr to 3.3×10¹⁵ cm⁻² by using ion beam sputtering method resulted in negligible transmutation yield.

Summary

(5) Further experimental work is necessary to confirm the phenomena;

-Dependence on the density of the atoms to be transmuted, the areal density of the atoms, the flux (flow rate) of D, the sample structure,

-Effect of surface contamination.

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